SAFETY DATA SHEET



1. Identification **SULPHUR TRIOXIDE**

Product identifier

130000031691 Other means of identification

SDS number Sulfuric Acid Anhydride * SO3 * Sulfur Trioxide, Stabilized * Sulfur Trioxide, Unstabilized **Synonyms** Sulfonating agent in surfactants, soaps, detergents and shampoo products. Flame retardants.

Polymer Additive. Manufacture of inorganic base chemicals. Drilling applications. Recommended use

Not to be used as a biocidal product. Not to be used as a drain cleaner. Not to be used as a direct

component of a cleaning product. Not to be used for cleaning sludge out of oil tanks. Recommended restrictions

Manufacturer/Importer/Supplier/Distributor information

Manufacturer

Company name Nexpera LLC

Address 131 Continental Dr. Suite 300

> Newark, DE 19713 United States of America

Website Nexpera.com **Telephone** 1-800-441-9362

1.4. Emergency telephone

number

CHEMTREC: 1-800-424-9300 (outside the U.S. 1-703-527-3887)

2. Hazard(s) identification

Physical hazards Corrosive to metals Category 1 **Health hazards** Category 1 Acute toxicity, inhalation Skin corrosion/irritation Category 1A

Category 1 Serious eye damage/eye irritation Specific target organ toxicity, single exposure Category 3 respiratory tract irritation

Specific target organ toxicity, repeated

Category 2 (respiratory system, teeth)

exposure (inhalation)

OSHA defined hazards Not classified.

Label elements



Signal word Danger

May be corrosive to metals. Fatal if inhaled. Causes severe skin burns and eye damage. May **Hazard statement**

cause respiratory irritation. May cause damage to organs (respiratory system, teeth) through

prolonged or repeated exposure by inhalation.

Precautionary statement

Prevention Keep only in original container. Do not breathe mist/vapors. Wash thoroughly after handling. Use

only outdoors or in a well-ventilated area. Wear protective gloves/protective clothing/eye

protection/face protection. Wear respiratory protection.

If swallowed: Rinse mouth. Do NOT induce vomiting. If on skin (or hair): Take off immediately all Response

contaminated clothing. Rinse skin with water/shower. Wash contaminated clothing before reuse. If inhaled: Remove person to fresh air and keep comfortable for breathing. If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do.

Continue rinsing.

Store in a well-ventilated place. Keep container tightly closed. Store locked up. Store in corrosive Storage

resistant container with a resistant inner liner.

Disposal Dispose of contents/container in accordance with local/regional/national/international regulations.

SULPHUR TRIOXIDE SDS US 1/8

937480 Version #: 03 Revision date: 08-November-2023 Issue date: 21-July-2017 Hazard(s) not otherwise classified (HNOC)

Reacts violently with water. Reacts with most metals to form flammable hydrogen gas.

Supplemental information

Corrosive to the respiratory tract.

3. Composition/information on ingredients

Substances

Chemical name	Common name and synonyms	CAS number	%
Sulfur trioxide		7446-11-9	100

Composition comments

All concentrations are in percent by weight unless ingredient is a gas. Gas concentrations are in percent by volume.

4. First-aid measures

Inhalation

Remove victim to fresh air and keep at rest in a position comfortable for breathing. Oxygen or artificial respiration if needed. Call a physician or poison control center immediately.

Skin contact

Take off immediately all contaminated clothing. Flush thoroughly with water for at least 15 minutes. Call a physician or poison control center immediately. Apply compresses of ice water while patient is being transported to medical facilities. Chemical burns must be treated by a physician. Wash contaminated clothing before reuse.

Eye contact

Immediately flush eyes with plenty of water for at least 15 minutes. Remove contact lenses, if present and easy to do. Continue rinsing. Call a physician or poison control center immediately.

Ingestion

Call a physician or poison control center immediately. Rinse mouth. Do not induce vomiting. If vomiting occurs, keep head low so that stomach content doesn't get into the lungs. Do not use mouth-to-mouth method if victim ingested the substance. Induce artificial respiration with the aid of a pocket mask equipped with a one-way valve or other proper respiratory medical device.

Most important symptoms/effects, acute and delayed

Burning pain and severe corrosive skin damage. Causes serious eye damage. Symptoms may include stinging, tearing, redness, swelling, and blurred vision. Permanent eye damage including blindness could result. May cause respiratory irritation. Sulfur trioxide forms sulfuric acid when exposed to water or moisture in the air. Prolonged exposure to sulfuric acid mist may cause cancer.

Indication of immediate medical attention and special treatment needed

Provide general supportive measures and treat symptomatically. Chemical burns: Flush with water immediately. While flushing, remove clothes which do not adhere to affected area. Call an ambulance. Continue flushing during transport to hospital. Keep victim warm. Keep victim under observation. Symptoms may be delayed.

General information

IF exposed or concerned: Get medical advice/attention. Take off all contaminated clothing immediately. Wash contaminated clothing before re-use. If you feel unwell, seek medical advice (show the label where possible). Ensure that medical personnel are aware of the material(s) involved, and take precautions to protect themselves. Show this safety data sheet to the doctor in attendance.

5. Fire-fighting measures

Suitable extinguishing media

Water fog. Foam. Powder. Carbon dioxide (CO2).

The product itself does not burn. Use fire-extinguishing media appropriate for surrounding materials. Carefully apply fine water mist or mid-expansion foam to slowly dilute to non fuming sulfuric acid. This process may release sulfuric acid mists into the air. Reaction with water and surrounding materials will generate heat and sulfuric acid mist.

Unsuitable extinguishing media

None known.

Specific hazards arising from the chemical

During fire, gases hazardous to health may be formed. Combustion products include: Sulfur

Special protective equipment and precautions for firefighters

Self-contained breathing apparatus and full protective clothing must be worn in case of fire.

Fire fighting equipment/instructions Specific methods

In case of fire and/or explosion do not breathe fumes. Move containers from fire area if you can do so without risk. In the event of fire, cool tanks with water spray. Do not get water inside container.

Use standard firefighting procedures and consider the hazards of other involved materials.

6. Accidental release measures

Personal precautions, protective equipment and emergency procedures Keep unnecessary personnel away. Keep people away from and upwind of spill/leak. Wear appropriate protective equipment and clothing during clean-up. Do not breathe vapors or spray mist. Do not touch damaged containers or spilled material unless wearing appropriate protective clothing. Ensure adequate ventilation. Local authorities should be advised if significant spillages cannot be contained. For personal protection, see section 8 of the SDS.

SULPHUR TRIOXIDE SDS US

Methods and materials for containment and cleaning up

This product is miscible in water. Should not be released into the environment. Prevent entry into waterways, sewer, basements or confined areas.

Large Spills: Stop the flow of material, if this is without risk. Dike the spilled material, where this is possible. Dilute spill to non-fuming sulfuric acid (<100%) using a water fog or aqueous foam. Remove product with clean and dry vacuum truck or pump to storage/salvage vessel. Following product recovery, flush area with water. Neutralize with lime, soda ash or other alkali material.

Small Spills: Neutralize with lime, soda ash or other alkali material. Flush area with water. Clean surface thoroughly to remove residual contamination.

Retain all contaminated water for removal and treatment. Put material in suitable, covered, labeled containers. For waste disposal, see section 13 of the SDS.

Environmental precautions

Avoid release to the environment. Inform appropriate managerial or supervisory personnel of all environmental releases. Prevent further leakage or spillage if safe to do so. Do not contaminate water. Avoid discharge into drains, water courses or onto the ground.

7. Handling and storage

Precautions for safe handling

Obtain special instructions before use. Do not handle until all safety precautions have been read and understood. Do not taste or swallow. Do not breathe vapors or spray mist. Provide adequate ventilation. Do not get in eyes, on skin, or on clothing. Wash hands thoroughly after handling. Wash contaminated clothing before reuse. When using do not eat, drink or smoke. Should be handled in closed systems, if possible. Use only outdoors or in a well-ventilated area. Do not mix water and/or aqueous solutions with sulfur trioxide. Wear appropriate personal protective equipment. Observe good industrial hygiene practices.

Conditions for safe storage, including any incompatibilities

Store locked up. Store in original tightly closed container. Store in a well-ventilated place. Store away from incompatible materials (see Section 10 of the SDS). Protect containers from damage. Never allow product to get in contact with water during storage. Store in a cool, dry place out of direct sunlight. Store in a corrosive resistant container.

8. Exposure controls/personal protection

Occupational exposure limits

NIOSH. Immediately Dangerous to Life or Health (IDLH) Values, as amended					
Material	Туре	Value			
Sulfur trioxide	IDLH	15 mg/m3			

Biological limit values

Appropriate engineering controls

No biological exposure limits noted for the ingredient(s).

Good general ventilation should be used. Ventilation rates should be matched to conditions. If applicable, use process enclosures, local exhaust ventilation, or other engineering controls to maintain airborne levels below recommended exposure limits. If exposure limits have not been established, maintain airborne levels to an acceptable level. Eye wash facilities and emergency shower must be available when handling this product.

Individual protection measures, such as personal protective equipment

Eye/face protection Wear chemical splash goggles in combination with a full-length face shield or an acid hood.

Skin protection

Hand protection Wear appropriate chemical resistant gloves. Be aware that the liquid may penetrate the gloves.

Frequent change is advisable.

Skin protection

Other Wear appropriate chemical resistant clothing. Full body chemical protective clothing. Chemical

resistant boots.

Respiratory protection Use a NIOSH-approved respirator as appropriate.

Thermal hazards None known.

General hygiene considerations

Observe any medical surveillance requirements. Always observe good personal hygiene measures, such as washing after handling the material and before eating, drinking, and/or smoking. Routinely wash work clothing and protective equipment to remove contaminants. Keep away from food and drink.

9. Physical and chemical properties

Appearance

Physical state Liquid.
Form Oily liquid.
Color Clear, colorless.

SULPHUR TRIOXIDE SDS US

937480 Version #: 03 Revision date: 08-November-2023 Issue date: 21-July-2017

Odor Acrid.

Odor threshold Not available.

рΗ < 1 (Aqueous solution)

62.24 °F (16.8 °C) / 144.14 °F (62.3 °C) (Alpha crystal) Melting point/freezing point

> 90.5 °F (32.5 °C) (Beta crystal) 62.24 °F (16.8 °C) (Gamma crystal)

Initial boiling point and boiling

range

112.64 °F (44.8 °C)

Flash point Does not flash

Evaporation rate < 1 (butyl acetate = 1)

Flammability (solid, gas) Not applicable. Upper/lower flammability or explosive limits

Explosive limit - lower (%) Not applicable. Not applicable. Explosive limit - upper (%)

510 mm Hg (37.7 °C / 99.9 °F) Vapor pressure

280 mm Hg (25 °C / 77°F)

Vapor density 2.8 (Air = 1)

1.84 (37.8 °C / 100 °F) Relative density

Solubility(ies)

Completely soluble. Reacts violently with water liberating sulfuric acid mist cloud. Solubility (water)

Partition coefficient -2.2

(n-octanol/water)

Not applicable. **Auto-ignition temperature Decomposition temperature** Not available. **Viscosity** Not available.

Other information

Density 0.63 g/cm3

Dynamic viscosity 6.62 mPa.s (104 °F (40 °C))

Explosive properties Not explosive.

Kinematic viscosity 10.5 mm²/s estimated

O3-S Molecular formula 80.06 g/mol Molecular weight Oxidizing properties Not oxidizing.

51.7 mN/m (122 °F (50 °C)) **Surface tension**

10. Stability and reactivity

Reactivity Reacts violently with water liberating sulfuric acid mist cloud. Contact with metal may release

flammable hydrogen gas. Reacts violently with strong alkaline substances. This product may react

with reducing agents. Can react with moisture in air to produce sulfuric acid mist.

Chemical stability Material is stable under normal conditions.

Possibility of hazardous reactions

Hazardous polymerization does not occur.

Conditions to avoid Heat. Contact with incompatible materials. Do not mix with other chemicals.

Water. Metals. Organic material. Nitrates. Chlorates. Perchlorates. Picrates. Carbides. Strong Incompatible materials

oxidizers. Reducing agents. Cyanides. Sulfides. Bases.

Hazardous decomposition

Version #: 03

products

937480

Sulfur oxides.

11. Toxicological information

Information on likely routes of exposure

Corrosive to the respiratory tract. Sulfuric acid is corrosive to eyes, skin and mucous membranes. Inhalation

Prolonged or repeated inhalation may damage respiratory tissue and may erode tooth enamel.

Causes severe skin burns. May be harmful in contact with skin. Skin contact

Eye contact Causes serious eye damage.

Revision date: 08-November-2023

SULPHUR TRIOXIDE SDS US Issue date: 21-July-2017

Fatal if swallowed. Ingestion

Symptoms related to the physical, chemical and toxicological characteristics Burning pain and severe corrosive skin damage. Causes serious eye damage. Symptoms may include stinging, tearing, redness, swelling, and blurred vision. Permanent eye damage including blindness could result. May cause respiratory irritation. Sulfur trioxide forms sulfuric acid when exposed to water or moisture in the air. Prolonged exposure to strong inorganic acid mists may cause cancer.

Information on toxicological effects

Acute toxicity Fatal if inhaled.

Components **Species Test Results**

Sulfur trioxide (CAS 7446-11-9)

Acute Inhalation Aerosol

LC50 Rat 0.375 mg/l, 4 Hours

Skin corrosion/irritation Causes severe skin burns. Serious eye damage/eye Causes serious eye damage. irritation

Respiratory or skin sensitization

Respiratory sensitization Not a respiratory sensitizer.

Skin sensitization This product is not expected to cause skin sensitization.

No data available to indicate product or any components present at greater than 0.1% are Germ cell mutagenicity

mutagenic or genotoxic.

The International Agency for Research on Cancer (IARC) has classified "strong inorganic acid Carcinogenicity

mists containing sulfuric acid" as a known human carcinogen, (IARC category 1). This

classification applies only to mists containing sulfuric acid and not to sulfuric acid or sulfuric acid

solutions.

IARC Monographs. Overall Evaluation of Carcinogenicity

Strong inorganic acid mists containing sulfuric acid 1 Carcinogenic to humans.

(CAS N/A)

NTP Report on Carcinogens

Strong inorganic acid mists containing sulfuric acid Known To Be Human Carcinogen.

(CAS N/A)

OSHA Specifically Regulated Substances (29 CFR 1910.1001-1053)

Not listed.

Reproductive toxicity This product is not expected to cause reproductive or developmental effects.

Specific target organ toxicity -

single exposure

May cause respiratory irritation.

Specific target organ toxicity repeated exposure

Sulfuric acid is corrosive to eyes, skin and mucous membranes. Prolonged or repeated inhalation may damage respiratory tissue and may erode tooth enamel.

Aspiration hazard Not an aspiration hazard.

12. Ecological information

Because of the low pH of this product, it would be expected to produce significant ecotoxicity upon **Ecotoxicity**

exposure to aquatic organisms and aquatic systems.

Persistence and degradability Not applicable.

The product is not expected to bioaccumulate. Bioaccumulative potential

Partition coefficient n-octanol / water (log Kow)

-2.2

937480

Mobility in soil The product is miscible with water.

Other adverse effects Sulfur trioxide reacts violently with water to form sulfuric acid and sulfuric acid mists.

13. Disposal considerations

Version #: 03

Disposal instructions Collect and reclaim or dispose in sealed containers at licensed waste disposal site. Do not allow

this material to drain into sewers/water supplies. Dispose of contents/container in accordance with

local/regional/national/international regulations.

Local disposal regulations Dispose in accordance with all applicable regulations.

Revision date: 08-November-2023

SULPHUR TRIOXIDE SDS US Issue date: 21-July-2017

Hazardous waste code D002: Waste Corrosive material [pH ≤2 or =>12.5, or corrosive to steel]

The waste code should be assigned in discussion between the user, the producer and the waste

disposal company.

Waste from residues / unused

products

Dispose in accordance with local regulations. Empty containers or liners may retain some product residues. This material and its container must be disposed of in a safe manner (see: Disposal

instructions).

Contaminated packaging Since emptied containers may retain product residue, follow label warnings even after container is

emptied. Empty containers should be taken to an approved waste handling site for recycling or

disposal.

14. Transport information

DOT

UN1829 **UN** number

UN proper shipping name

Sulfur trioxide, stabilized (RQ = 100 LBS)

Transport hazard class(es)

Class 8 **Subsidiary risk** 6.1 Label(s) 8, 6.1 Packing group

Environmental hazards

Marine pollutant No.

Special precautions for user Read safety instructions, SDS and emergency procedures before handling. Special provisions 2, 387, B9, B14, B32, B49, B77, N34, T20, TP4, TP13, TP25, TP26, TP38, TP45

Packaging exceptions None Packaging non bulk 227 Packaging bulk 244

IATA

UN number UN1829

FORBIDDEN DANGEROUS GOODS **UN proper shipping name**

Transport hazard class(es)

Class 8 **Subsidiary risk**

Not applicable. Packing group

Environmental hazards No. **ERG Code**

Special precautions for user Read safety instructions, SDS and emergency procedures before handling.

IMDG

UN number UN1829

UN proper shipping name Transport hazard class(es) SULPHUR TRIOXIDE, STABILIZED

Class 8 Subsidiary risk

Packing group **Environmental hazards**

Marine pollutant No. **EmS** F-A. S-B

Special precautions for user Read safety instructions, SDS and emergency procedures before handling.

Transport in bulk according to Annex II of MARPOL 73/78 and

Not established. However, this product is a liquid and if transported in bulk covered under

MARPOL 73/78, Annex I.

the IBC Code

15. Regulatory information

US federal regulations This product is a "Hazardous Chemical" as defined by the OSHA Hazard Communication

Standard, 29 CFR 1910.1200.

TSCA Section 12(b) Export Notification (40 CFR 707, Subpt. D)

Not regulated.

CERCLA Hazardous Substance List (40 CFR 302.4)

SARA 304 Emergency release notification

Sulfur trioxide (CAS 7446-11-9) 100 LBS

SULPHUR TRIOXIDE SDS US

OSHA Specifically Regulated Substances (29 CFR 1910.1001-1053)

Not listed.

Toxic Substances Control Act (TSCA)

All components of the mixture on the TSCA 8(b) inventory are designated

"active".

Superfund Amendments and Reauthorization Act of 1986 (SARA)

SARA 302 Extremely hazardous substance

Chemical name	CAS number	Reportable quantity (pounds)	Threshold planning quantity (pounds)	Threshold planning quantity, lower value (pounds)	Threshold planning quantity, upper value (pounds)
Sulfur trioxide	7446-11-9	100	100		

SARA 311/312 Hazardous

Yes

chemical

Classified hazard categories

Corrosive to metal

Acute toxicity (any route of exposure)

Skin corrosion or irritation

Serious eye damage or eye irritation

Specific target organ toxicity (single or repeated exposure)

Hazard not otherwise classified (HNOC)

SARA 313 (TRI reporting)

Chemical name CAS number % by wt. Sulfur trioxide 7446-11-9 100

Other federal regulations

Clean Air Act (CAA) Section 112 Hazardous Air Pollutants (HAPs) List

Not regulated.

Clean Air Act (CAA) Section 112(r) Accidental Release Prevention (40 CFR 68.130)

Sulfur trioxide (CAS 7446-11-9)

Clean Water Act (CWA)

Hazardous substance

Section 112(r) (40 CFR

68.130)

Safe Drinking Water Act

Not regulated.

(SDWA)

Drug Enforcement Administration (DEA). List 2, Essential Chemicals (21 CFR 1310.02(b) and 1310.04(f)(2) and **Chemical Code Number**

Sulfur trioxide (CAS 7446-11-9) 6552

Drug Enforcement Administration (DEA). List 1 & 2 Exempt Chemical Mixtures (21 CFR 1310.12(c))

Sulfur trioxide (CAS 7446-11-9) 20 %WV

DEA Exempt Chemical Mixtures Code Number

Sulfur trioxide (CAS 7446-11-9) 6552

Food and Drug Total food additive Administration (FDA) Direct food additive GRAS food additive

US state regulations

US. Massachusetts RTK - Substance List

Sulfur trioxide (CAS 7446-11-9)

US. New Jersey Worker and Community Right-to-Know Act

Sulfur trioxide (CAS 7446-11-9)

US. Pennsylvania Worker and Community Right-to-Know Law

Sulfur trioxide (CAS 7446-11-9)

US. Rhode Island RTK

Sulfur trioxide (CAS 7446-11-9)

California Proposition 65



WARNING: This product can expose you to chemicals including Strong inorganic acid mists containing sulfuric acid, which is known to the State of California to cause cancer. For more information go

to www.P65Warnings.ca.gov.

California Proposition 65 - CRT: Listed date/Carcinogenic substance

Strong inorganic acid mists containing sulfuric acid Listed: March 14, 2003

(CAS N/A)

SULPHUR TRIOXIDE SDS US 7/8 937480 Version #: 03 Revision date: 08-November-2023 Issue date: 21-July-2017

International Inventories

New Zealand

Philippines

Country(s) or region	Inventory name	On inventory (yes/no)*
Australia	Australian Inventory of Industrial Chemicals (AICIS)	Yes
Canada	Domestic Substances List (DSL)	Yes
Canada	Non-Domestic Substances List (NDSL)	No
China	Inventory of Existing Chemical Substances in China (IECSC)	Yes
Europe	European Inventory of Existing Commercial Chemical Substances (EINECS)	Yes
Europe	European List of Notified Chemical Substances (ELINCS)	No
Japan	Inventory of Existing and New Chemical Substances (ENCS)	Yes
Korea	Existing Chemicals List (ECL)	Yes

Philippine Inventory of Chemicals and Chemical Substances

(PICCS)

Taiwan Taiwan Chemical Substance Inventory (TCSI) Yes Toxic Substances Control Act (TSCA) Inventory United States & Puerto Rico Yes

16. Other information, including date of preparation or last revision

New Zealand Inventory

Issue date 21-July-2017 **Revision date** 08-November-2023

Version # 03

NFPA ratings



Revision date: 08-November-2023

Disclaimer

Nexpera LLC cannot anticipate all conditions under which this information and its product, or the products of other manufacturers in combination with its product, may be used. It is the user's responsibility to ensure safe conditions for handling, storage and disposal of the product, and to assume liability for loss, injury, damage or expense due to improper use. The information in the sheet was written based on the best knowledge and experience currently available.

Issue date: 21-July-2017

Version #: 03

937480

Yes

Yes

^{*}A "Yes" indicates that all components of this product comply with the inventory requirements administered by the governing country(s) A "No" indicates that one or more components of the product are not listed or exempt from listing on the inventory administered by the governing country(s).